國立清華大學命題紙

 The tungsten carben complex 1 in the presence of GaBr₃ catalyzes the ring-opening polymerization of norbornene 2. (5%)

propose a mechanism for this catalytic reaction.

- 2. (C₅H₅)Fe(CO)₂H eliminates a colorless gas A, forming a purple-brown solid B with the empirical formula C₇H₅O₂Fe. Reaction of B with I₂ gives a brown solid C of empirical formula C₇H₅O₂FeI. Treatment of C with Tl(C₅H₅) affords a solid D of formula C₁₂H₁₀O₂Fe, which gives off a colorless gas and produces an organge solid E of formula C₁₀H₁₀Fe. Propose structures for A~E. (5%)
- Predict the structures of the reaction products.

(a) OC
$$\frac{CH_3}{CO}$$
 + $P(CH_3)Ph_2$ (2%)

(no gas evolution)

(b)
$$Ir$$
 PPh_3 Ph_3P CO PPh_3 (2%)

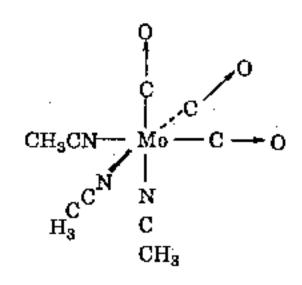
4. $Co(NH_3)_6^{3+}$ is reduced slowly by $Cr(H_2O)_6^{2+}$ (k = 8.9x10⁻⁵M⁻¹S⁻¹), but A is reduced much more rapidly by $Cr(H_2O)_6^{2+}$ (k=17.6 M⁻¹S⁻¹). What are the mechanisms for these redox reactions and predict the structures of the final products. (5%)

$$\left[\begin{array}{c} (NH_3)_5C_0-\sqrt{\begin{array}{c} \\ \\ \end{array}} - \begin{bmatrix} 0 \\ 1 \\ C \\ NH_2 \end{array}\right]^{3+}$$

- 5. Draw all the possible isomers for Co(en)₂Cl₂+ and assign absolute configurations for the isomers. (6%)
- If one does not take into account Jahn-Teller effect, please show spectrum terms and possible d-d transition bands for the giving complexes below. (9%)
 (a) CrF₆³⁻ (weak-field) (b) NiF₄²⁻ (c) Co(H₂O)₆²⁺ (weak-field)
- 7. Of the complexes below, which will show the strongest absorption intensities in d-d transitions? Give your answer. (3%)

$$MnF_4^{2-}$$
, $Mn(H_2O)_6^{2+}$, NiF_4^{2-} , $Ni(H_2O)_6^{2+}$

8. Determine the number of IR active C-O stretching modes for fac-Mo(CO)3(CH3CN)3. Show clearly your calculation procedure. (5%)



$\mathbf{C_{3v}}$	E	$2C_3$	$3\sigma_{\!\scriptscriptstyle V}$		
$\mathbf{A_1}$	1	1	Į	z	x^2+y^2,z^2
$\mathbf{A_2}$		1			
E	2	-1	0	(x,y), (R _x , R _y)	(x ² -y ² ,xy),(xz,yz)

- 9. Show the structure and point groups for (Me₃Si)₃N. Why is the structure different from that of (Me₃C)₃N? Give your reason. (3%)
- 10. Which do you expect to be more acidic? Give your answer. (3%)

11. Explain the term "proton affinity". Give the a clear definition. (2%)

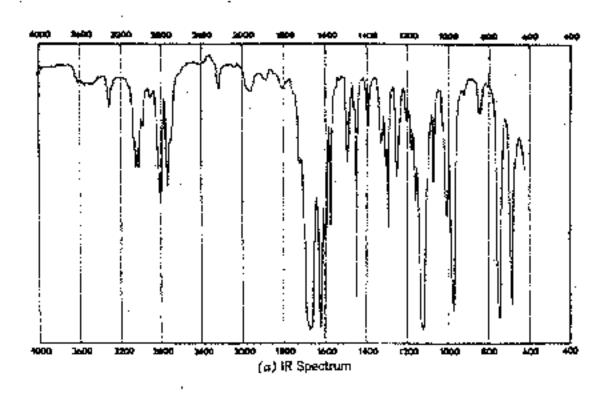
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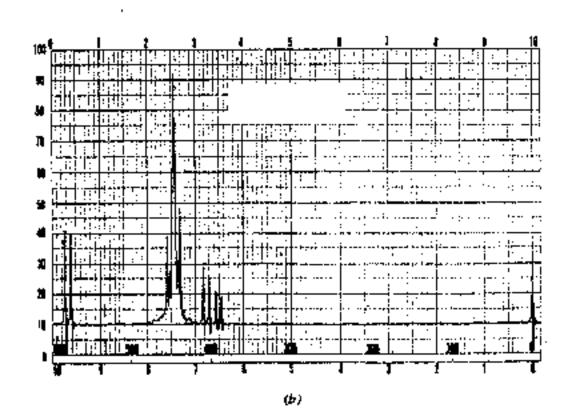
12. Give the structures of the major products for the following transformations, and propose a mechanism for the Cope elimination of compound D. (6%)

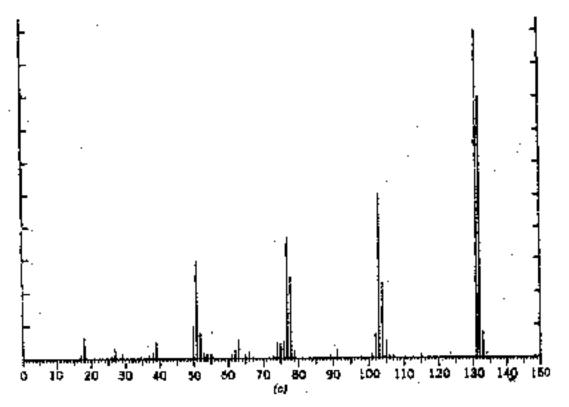
- 13. How many hydrogens are replaced by deuterium when each of the following compounds is treated with NaOD in D₂O? Write an equation for the reaction. (4%)
 - (a) Cyclohexanecarbaldehyde
 - (b) 3-methylcyclopentadecanone (muscone)
- 14. Muscone (I) (Problem 13-b), an odorous principle of natural musk from the musk deer, is optically active and has (R)-(-)- form. It is heated with hydrazine hydrate and sodium hydroxide in diethylene glycol to give compound II, which is optically inactive. Write the structures of compounds I and II and show the absolute configuration for I, and also explain why compound I is optically active and II is inactive. (4%)
- 15. Compound X (C7H₁₂) reacts with dry HCl at -20°C to give Y (C7H₁₃Cl), which reacts with potassium t-butoxide in t-butyl alcohol to give a small amount of X and mainly Z (C7H₁₂). Ozonolysis of Z gives cyclohexanone and formaldehyde. What are compounds X through Z? (6%)

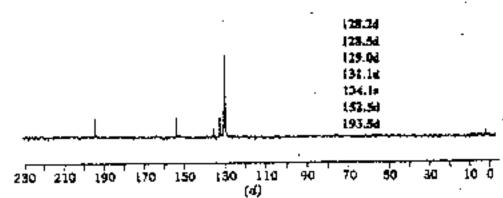
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16. Give the structure of a compound having a molecular weight of 132.0575, and UV λ_{max} = 248 nm (ϵ = 15000) on the basis of its IR, PMR, Mass and ¹³C NMR spectra shown below. Please indicate the possible assignments. (5%)









17. Show how you would accomplish each of the following multistep synthetic transformations in good yield. (12%)

(b)
$$CO_2H$$
 ——— $(CH_3)_2CHCH_2$ — NH_2

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18. Give a stepwise arrow pushing mechanism for the following transformation (3%)

19. Give the structure of major product(s), clearly indicate the stereochemistry if nocessary, or provide necessary reagent(s) to complete each of the following transformations. (10%)

(d) S BuLi 1 eq. CH₃I
$$Hg^{2+}$$
, H₂O ?