八十六學年度
 生命科學
 系(所)
 内
 組領土班研究生入學者試

 科目
 近代物理
 科院
 1102 共 2 頁第 / 頁 *請在試卷【答案卷】內作答

- 1. A spin system consists of n_1 and n_2 particles whose energy levels are given by E_1 and E_2 . Please find the temperature of the system and explain how it is possible to reach negative temperature. (10%)
- 2. Design an experiment to measure osmotic pressure of a sugar solution in a semi-permeable membrane. (10%)
- 3. Consider a gas of uniform and constant density and temperature with an average velocity given by

$$U_x = A + By$$

$$U_y = U_z = 0$$

Let F be the friction force across a plane perpendidular to the y axis, please find its viscosity. (10%)

- 4. Define and give specific example for the phase transition of the second kind and of the first kind (10%)
- 5. Describe Stern-Gertach experiment, normal and anomalous Zeeman effect, and parity violation (10%)
- Describe physical process responsible for Rayleigh scattering and multiatomic resonance fluorescence. Please also give examples of natural phenomena involving these processes (10%).
- 7. Use the uncertainty principle to estimate the ground-state energy for a particle of mass m moving in the potential well U(x) = k|x|, where k > 0 (10%).
- 8. Prove that the quantity τ defined by $\tau = t_{1/2} / \ln 2$ is the mean life time of the decaying nuclei (10%).
- 9. Suppose that $v = 5.0 \times 10^{14} \text{ Hz}$ and T = 3000 K. According to Planck, what is the average energy of the oscillator? (10%).
- 10. According to a simple model, the energy of a free electron in a crystal is the following function of the wave number: E = A · Bcos ka where A and B are constants, and a is the distance between adjacent atoms. What value of the effective mass can you deduce from this formular (10%).

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Gravitational constant Speed of light Electronic charge Planck's constant

Fine-structure constant
Electron mass
Muon mass
Pion mass
Proton mass
Bohr radius
Classical electron radius
Bohr magneton
Nuclear magneton
Boltzmann constant
Stephan-Boltzmann constant

Avogadro's number Gas constant

Proton magnetic moment Neutron magnetic moment $G = 6.67 \times 10^{-4} \text{ dyn cm}^2 \text{ gm}^{-4}$ $c = 3 \times 10^{10} \text{ cm sec}^{-1}$ $e = 4.8 \times 10^{-10} \text{ esu} = 1.6 \times 10^{-10} \text{ C}$ $h = 6.625 \times 10^{-10} \text{ erg sec}$ $h = h/2\pi = 1.05 \times 10^{-10} \text{ erg sec}$ $\alpha = e^2/hc = 1/137$ $m_e = 9.1 \times 10^{-10} \text{ gm} = 0.51 \text{ MeV}$ $m(\mu) = 207 m_e = 106 \text{ MeV}$

 $m(\pi) = 270 \text{ m}, = 140 \text{ MeV}$ $M = 1.6 \times 10^{-24} \text{ gm} = 938 \text{ MeV}$ $a_0 = \hbar^4/mc^2 = 0.53 \times 10^{-6} \text{ cm}$ $r_1 = e^2/mc^2 = 2.82 \times 10^{-12} \text{ cm}$ $\mu_2 = e\hbar/2m_e c = 0.927 \times 10^{-22} \text{ erg gauss}^{-1}$ $\mu_3 = e\hbar/2Mc = 0.505 \times 10^{-22} \text{ erg gauss}^{-1}$ $k = 1.38 \times 10^{-12} \text{ erg (°K)}^{-1}$ $\sigma = (\pi^2/80)(k^2/\hbar^2c^2)$ $= 5.7 \times 10^{-6} \text{ erg cm}^{-1} \text{ sec}^{-1} \times (°K)^{-6}$ $N_4 = 6.02 \times 10^{22} \text{ mole}^{-1}$

 $R = N_0 k = 2 \text{ cal-mole}^{-1.0} \text{K}^{-1}$ = 8.3 × 10^t ergs mole^{-1.0} K⁻¹ $\mu_t = 2.79 \text{ nm}$ $\mu_n = -1.01 \text{ nm}$