國立清華大學命題紙

98 學年度 工學院分子工程學程 碩士班入學考試

科目__普通物理及普通化學__ 科目代碼__0902__共__1__頁第__1__頁 *請在【答案卷卡】內作答

General Chemistry

- 1. Four reagents were mixed. The solutions mixed together are 150.0 mL of 0.100 M hydrochloric acid, 100.0 mL of 0.200 M of nitric acid, 500.0 mL of 0.0100M calcium hydroxide, and 200.0 mL of 0.100 M rubidium hydroxide. Is the resulting solution neutral? If not, calculate the concentration of excess H⁺ or OH⁻ ions left in solution. [8%]
- 2. For the reaction

$$P_{4(g)} \longrightarrow 2P_{2(g)}$$

At 520 K and a pressure of 128 torr an equilibrium mixture P_4 and P_2 has a density of 0.4 g/L. Calculate K_p for the above reaction at 520 K. (Atomic mass of P is 31 g/mol.) [8%]

3. The ionic product [H⁺][OH⁻], which is the equilibrium constant for the dissociation of water,

$$H_2O \rightleftharpoons H^+ + OH^-$$

is $1.00 \times 10^{-14} \text{ mol}^2 \text{dm}^{-6}$ at 25.0 °C and $1.45 \times 10^{-14} \text{ mol}^2 \text{dm}^{-6}$ at 30.0 °C.

- (a) Deduce ΔH° and ΔS° for the process. [5%]
- (b) Calculate the value of the ionic product at body temperature (37 °C). [5%]
- 4. What is the maximum number of electrons in an atom that have these quantum numbers? [6%]
 - (a) n = 4, $m_s = +1/2$
 - (b) n = 2, l = 2
 - (c) $n = 1, l = 0, m_l = 0$
- 5. Predict the molecular structure and the bond angles for each of the following. [6%]
 - (a) ICl₃
 - (b) TeF₄
 - (c) PCl₅
- 6. Which has the lowest (ground-state) energy, an electron trapped in a one-dimensional box of length 10⁻⁶ m or 10⁻⁸ m? Why? [4%]
- 7. Rearrange ionization energies of the following gas species in the order from large to small. [4%]
 - (a) $Si_{(g)} \to Si^+_{(g)} + e^-$
 - (b) $Ar_{(g)} \rightarrow Ar_{(g)}^{+} + e^{-}$
 - (c) $P^{+}_{(g)} \to P^{+2}_{(g)} + e^{-}$
 - (d) $Br_{(g)} \rightarrow Br_{(g)}^{+} + e^{-}$
- 8. Rearrange the following lattice energy values in the order from large to small. [4%]
 - (a) CaSe, (b) Na₂Se, (c) CaTe, (d) Na₂Te

General Physics

- 9. Viewed from a large scale (such as in satellite images), typhoons attacking Taiwan always swirl in a counterclockwise manner, Explain why. [10%]
- 10. Explain the following terms. Be as specific as you can. [10%]
 - (a) Aberration; (b) Single-slit diffraction.
- 11. Explain the basic structure and the operating principle of (a) gratings and (b) beam splitters as frequently used components in spectroscopy. Be as specific as you can. [10%]
- 12. Under atmospheric pressure, the sound velocity σ in air is expected to vary with temperature T. Explain why and find the exponent n in the power-law dependence of $\sigma \propto T^n$. [10%]
- 13. Show that a pair of electric dipoles separated by a center-to-center distance r experience dipole-dipole interaction potential that is proportional to r^{-n} . Should the interaction attractive or repulsive? What is the value of n? [10%]